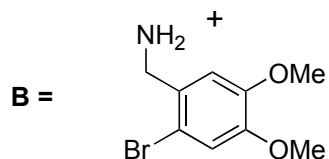
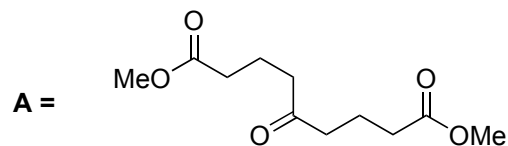
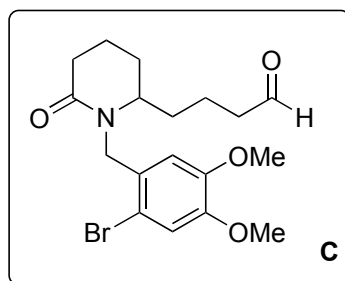


# Total Synthesis of (–)-Histrionicotoxin through a Stereoselective Radical Translocation–Cyclization Reaction

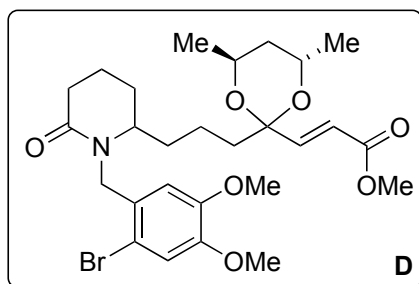
Sato, M.; Azuma, H.; Daigaku, A.; Sato, S.; Takasu, K.; Okano, K.; Tokuyama, H.; *Angew. Chem. Int. Ed.* **2017**, *56*, 1087–1091.



1–3

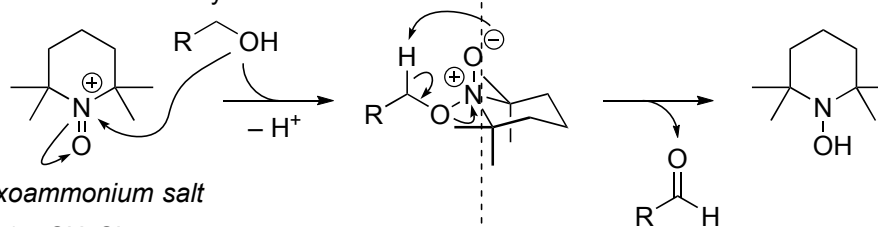


4–6



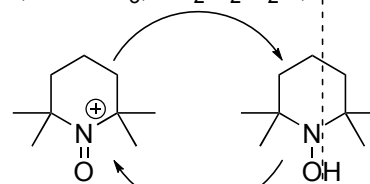
Step 3 mechanism:

Oxidation of alcohol to aldehyde



- 1) NaBH(OAc)<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt
- 2) K-Selectride, THF, 0 °C to rt
- 3) cat. TEMPO, NaOCl, KBr, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O, 0 °C

primary catalytic oxidant



TEMPO is oxidized to oxoammonium salt

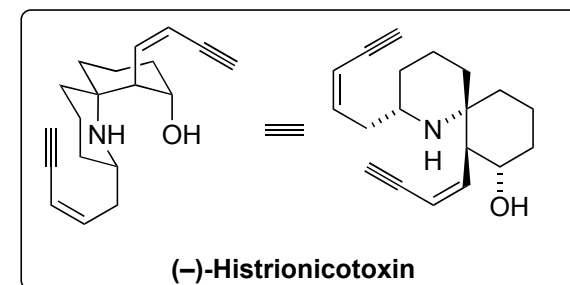
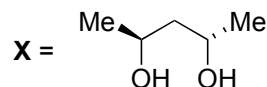
secondary catalytic oxidant



stoichiometric oxidant

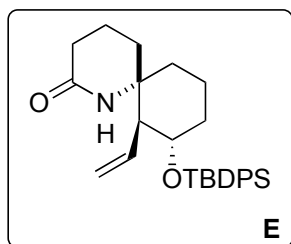


- 4) vinylmagnesium bromide, THF, –78 °C to 0 °C
- 5) methyl acrylate, cat. Grubbs II, CH<sub>2</sub>Cl<sub>2</sub>, reflux
- 6) MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt
- 7) diol **X**, cat. TMSOTf, TMSO*i*-Pr, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt

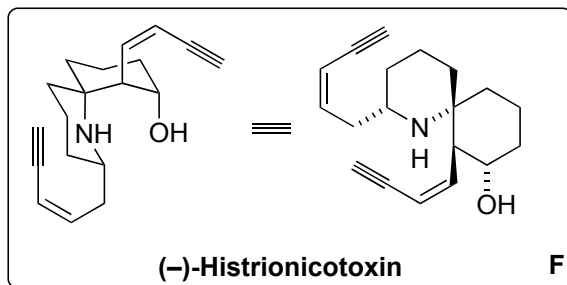




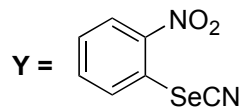
9–14



15–22



- 9) cat.  $\text{FeCl}_3$ ,  $\text{SiO}_2$ , acetone, rt
- 10)  $\text{Li}$ ,  $\text{NH}_3$ ,  $t\text{-BuOH/THF}$ ,  $-78^\circ\text{C}$ , 1 h
- 11)  $\text{PBU}_3$ , reagent **Y**, THF, 1 h



- 12)  $p\text{-TSA}$ , toluene, reflux
- 13)  $\text{TBDPSCl}$ , imidazole,  $\text{CH}_2\text{Cl}_2$ , rt
- 14)  $m\text{-CPBA}$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$  to rt

- 15)  $\text{Ti}(\text{O}i\text{-Pr})_4$ ,  $\text{Et}_2\text{SiH}_2$ , THF, reflux  
then cool to  $-78^\circ\text{C}$ ,  $\text{allylMgCl}$ , cat.  $\text{ZnCl}_2$
- 17) TBAF, THF
- 18)  $\text{SOCl}_2$ , imidazole,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$ , 1 h
- 19)  $\text{O}_3$ , then  $\text{Me}_2\text{S}$ ,  $\text{CH}_2\text{Cl}_2/\text{MeOH}$ ,  $-78^\circ\text{C}$  to rt
- 20)  $\text{PhP}_3\text{CH}_2\text{I}$ ,  $\text{KHMDS}$ , THF,  $-78^\circ\text{C}$
- 21) TMS-acetylene, cat.  $\text{Pd}(\text{PPh}_3)_4$ , cat.  $\text{CuI}$ ,  $\text{Et}_2\text{NH}$
- 22)  $\text{LiAlH}_4$ , THF,  $0^\circ\text{C}$  to rt

*Mechanism of step 8?*

*What other reagents other than  $(\text{TMS})_3\text{SiH}$  could be used? Advantages/disadvantages?*

*$n\text{Bu}_3\text{SnH}$  or  $\text{Ph}_3\text{SnH}$  or  $n\text{Bu}_3\text{GeH}$*

***Sn** = toxic, more reactive and lower diastereoselectivity*

***Ge** = more stable and less toxic, can sometimes improve yield in cyclization. expensive*

*Name Reaction?*

**Stork-Zhao reaction**

*Name 3 alternative methods for the formation of this functional group (also in other stereo- and regioselectivities)*

*terminal E- vinyl iodide:*

**Takai hydrozirconation/iodination**

*internal vinyl iodide:*

**Hoveyda Ni-catalyzed hydroalumination of alkynes**

**beta-oxido-phosphonium reaction from aldehydes and phosphonium salt**