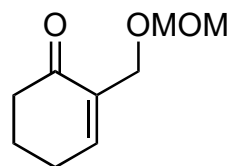


# Asymmetric Total Synthesis of Clionastatins A and B

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1-5



6-8

1) (*R*)-Me-CBS,  $\text{BH}_3 \cdot \text{Me}_2\text{S}$   
then  $\text{Ac}_2\text{O}$ , DMAP,  $\text{Et}_3\text{N}$

2) KHMDS,  $-78^\circ\text{C}$   
then TMSCl,  $23^\circ\text{C}$   
then  $80^\circ\text{C}$

3) NMM, *i*-BuOCOCI  
then  $(\text{Ph}_2\text{Te}_2, \text{DIBAL-H})$

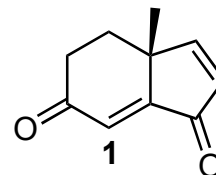
4) **1**,  $\text{Et}_3\text{B}$ , air  
then MeOH

5)  $\text{Et}_3\text{N}$ ,  $\text{Tf}_2\text{O}$

6)  $\text{NaBH}_4$ ,  $\text{CeCl}_3 \cdot 7\text{H}_2\text{O}$

7)  $\text{Pd}(\text{OAc})_2$ , dppp, PMP  
then DMP

8) 6M HCl, MeOH

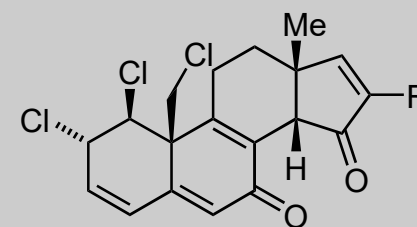


1) Name of the reaction

2) Name of the reaction

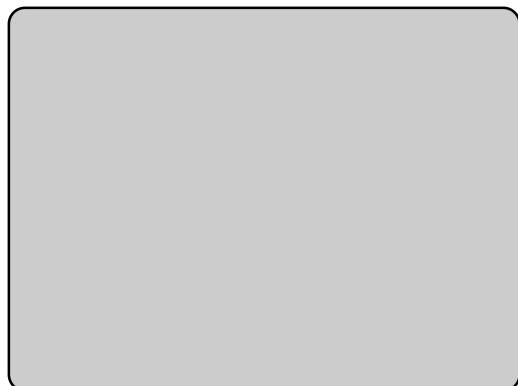
6) Hint: mixture of diols

7) Name of the reaction

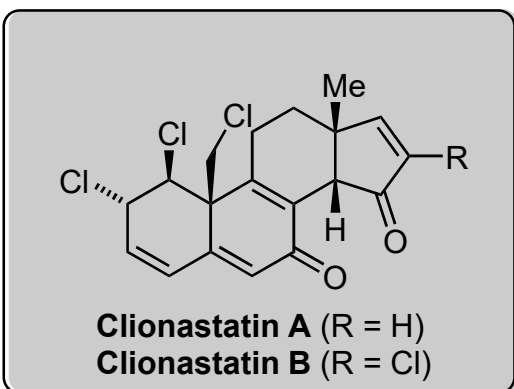


**Clionastatin A** ( $\text{R} = \text{H}$ )

**Clionastatin B** ( $\text{R} = \text{Cl}$ )



9-17  
↓



- 9)  $\text{SOCl}_2$ , DMF
- 10)  $\text{SeO}_2$ , dioxane
- 11)  $\text{Et}_4\text{NCl}_3$  (3.0 eq)
- 12)  $\text{Ac}_2\text{O}$ , DMAP,  $\text{Et}_3\text{N}$
- 13)  $\text{TiCl}_4$ , LiCl
- 14)  $\text{PPh}_3$ ,  $\text{CCl}_4$
- 15) IBX,  $\text{Ph}_2\text{Se}_2$   
*then* LiOH
- 16) Martin sulfurane
- 17)  $\text{Et}_4\text{NCl}_3$

9) Hint: Deoxychlorination does not occur; instead, a sultine is formed.  
Sultine: cyclic ester of a sulfinic acid.

11) Name of the reagent

16) Structure of Martin sulfurane

