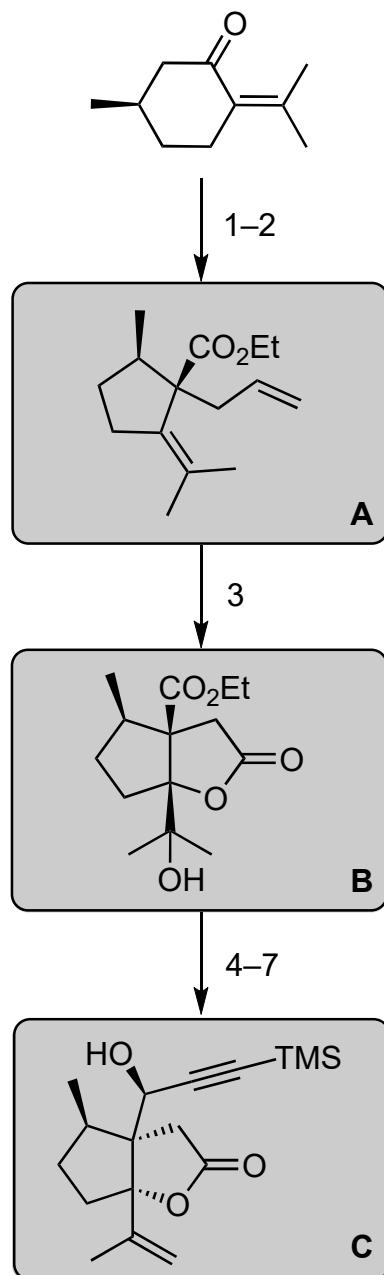


Site-Specific Photochemical Desaturation Enables Divergent Syntheses of *Illicium* Sesquiterpenes

Shen, Y.; Li, L.; Xiao, X.; Yang, S.; Hua, Y.; Wang, Y.; Zhang, Y.-W.; Zhang, Y.

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1) Br_2 , Et_2O ;
then NaOEt
2) KHMDS , HMPA , allyl iodide, THF , -78°C

3) $m\text{CPBA}$, NaHCO_3 , MeCN , 0°C ;
then RuCl_3 , NaIO_4 , r.t.

4) SOCl_2 , Et_3N , CH_2Cl_2 , -78°C
5) KHMDS ; then LiAlH_4
6) DMP
7) 1, THF , -78°C to 23°C

$\text{CIMg}\equiv\text{TMS}$
1

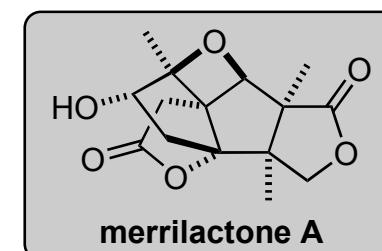
Name the starting material
(*R*)-pulegone

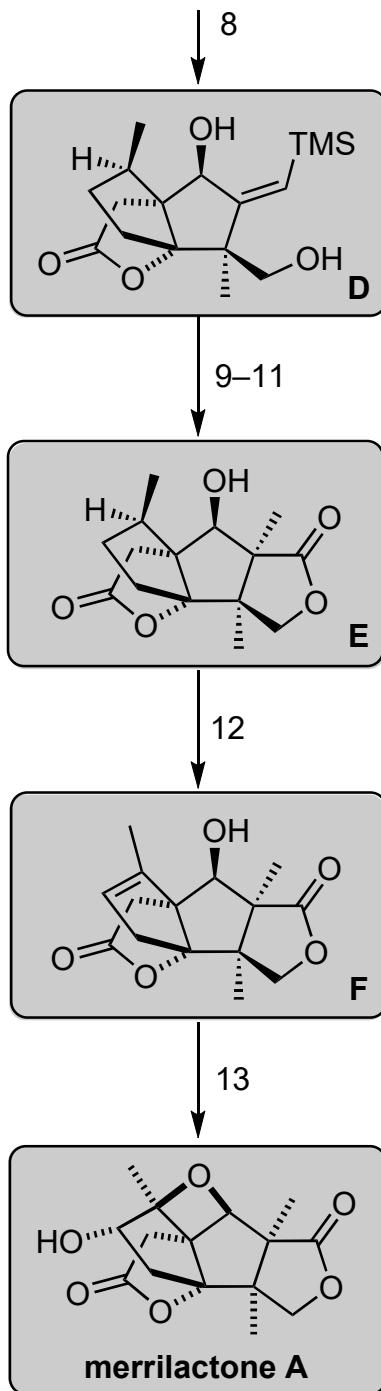
1) Please name the reaction and
describe the mechanism.
Favorskii rearrangement

3) *hint*: a heterocycle is formed

5) *hint*: sneaky trick to ensure
chemoselective reduction

7) Product obtained as single
diastereomer; suggest which
one and develop a rationale.





8) $\text{Pd}(\text{OAc})_2$, B_2pin_2 , toluene/MeOH, 50 °C;
then H_2O_2 , NaOH, THF, 0 °C

9) $p\text{-TsOH}$, MeCN, 23 °C
10) CDI, CH_2Cl_2 , 23 °C;
then PhSeNa
11) TTMSS, AIBN, benzene, 80 °C

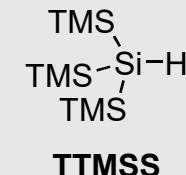
12) Ph_2CO , NBS, HFIP, violet LED

F is sometimes also referred to as *Danishefsky's intermediate*

13) $m\text{CPBA}$, EtOAc, 23 °C;
then $p\text{TsOH}\cdot\text{H}_2\text{O}$, EtOAc/MeOH, 23 °C

8) *hint:*
(a) product chemical formula:
 $\text{C}_{17}\text{H}_{28}\text{O}_4\text{Si}$

(b) a carbocycle and a quaternary center are formed

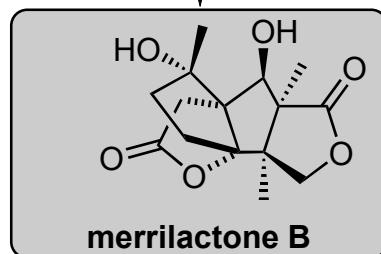


12) *hint:*
(a) look at the title
(b) the stereocenter that delivered the chiral information for the synthesis is planarized in this step

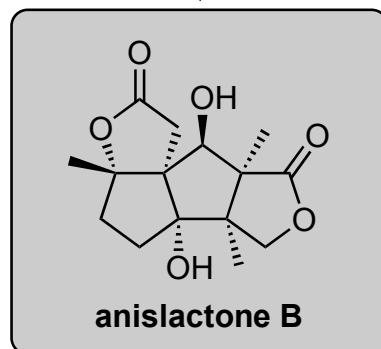
Additional:

F

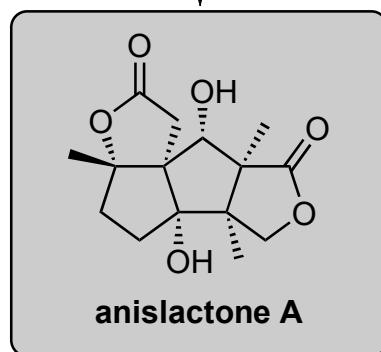
14



15



16



14) $\text{Ph}(i\text{PrO})\text{SiH}_2, \text{Co}(\text{acac})_2, \text{O}_2, \text{THF}, 23^\circ\text{C}$

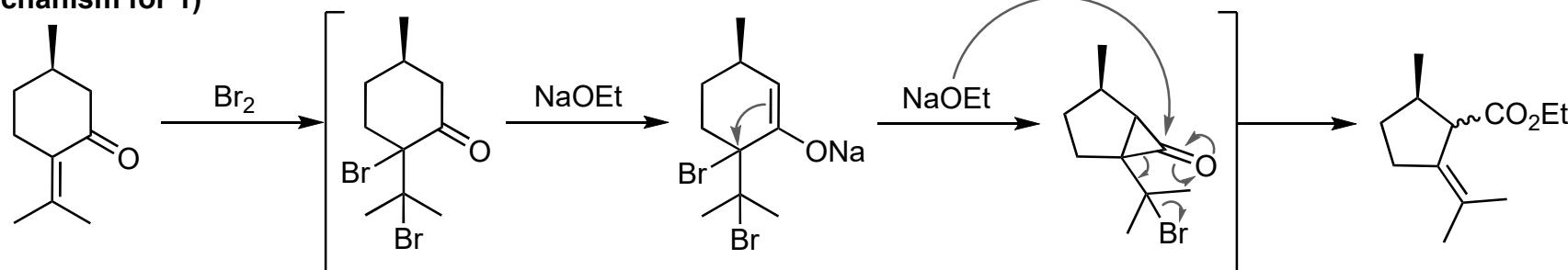
14) Name this reaction
Mukaiyama hydration

15) $\text{K}_2\text{CO}_3, \text{MeOH}, 23^\circ\text{C};$
then HCl

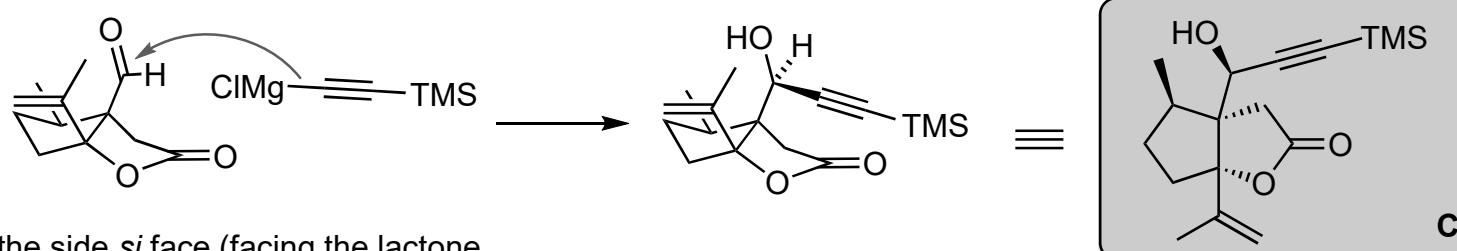
16) $\text{K}_2\text{CO}_3, \text{MeOH}, 23^\circ\text{C};$
then HCl

16) *hint: epimerization*

Mechanism for 1)



Potential stereochemical rationale for 7)



the side *si* face (facing the lactone ring) would be less congested in this reactive conformation

Plausible mechanistic pathways for 8)

J. Am. Chem. Soc. 2007, 129, 1874–1875.

