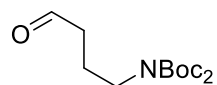


## Scalable Total Synthesis of Portimine A and B Reveals the Basis of Their Potent and Selective Anti-cancer Acitivity

J. Tang, W. Li, T.-Y. Chiu, Z. Luo, C. T. Chong, Q. Wei, F. Martinez-Peña, N. Gazaniga, Y. Y. See, L. L. Lairson, C. G. Parker, P. S. Baran

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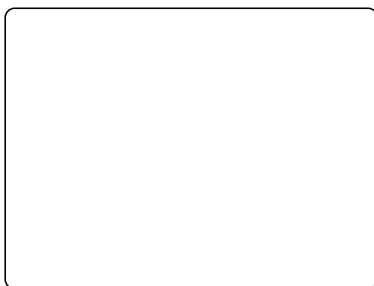


1 – 8



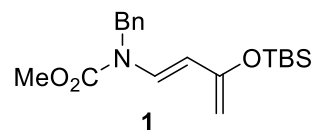
**A**

9 – 14

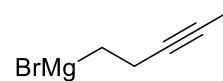


**B**

- HCHO, pyrrolidine, propionic acid, H<sub>2</sub>O, *i*-PrOH
- 1**, [Co(*t*-Bu-Salen)]SbF<sub>6</sub>, 4Å MS, DCM, r.t.
- NaBH<sub>4</sub>, MeOH, DCM, 0 °C
- TBAF, THF, r.t.
- TEMPO, NaOCl, KBr, NaHCO<sub>3</sub>, DCM, H<sub>2</sub>O, 0 °C
- 2**, THF, –78 °C
- DMP, NaHCO<sub>3</sub>, DCM, 0 °C
- TFA, DCM, r.t.

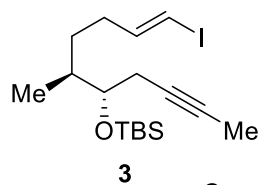


**1**

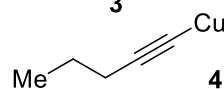


**2**

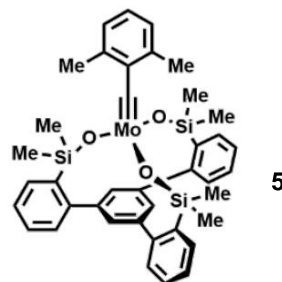
- 3**, *t*-BuLi, THF, –78 °C then **4**, *n*-Bu<sub>3</sub>P, THF, –78 °C then **A**, THF, –78 °C then Comins' rgt., THF, 0 °C
- TrocCl, DMAP, DCM, reflux
- 5**, 5Å MS, PhMe, 80 °C
- PTSA, DCM, MeOH, H<sub>2</sub>O, 50 °C
- XPhosAuNTf<sub>2</sub>, DCM, reflux
- Ru(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>, TBHP, TBAI, MeCN, PhMe, H<sub>2</sub>O, r.t.



**3**



**4**



**5**

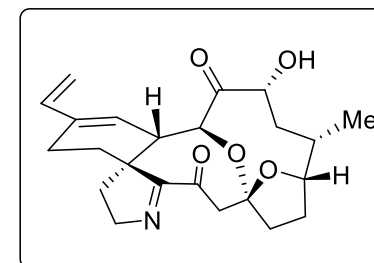
Step 2: Please, name the reaction, the reagent **1** and show structure of the salen ligand.

Step 5: Please, show the mechanism.

Step 11: Who developed this type of catalyst?

Hint step 12: Concomitant silyl deprotection.

Hint step 13: Three heterocycles are formed.

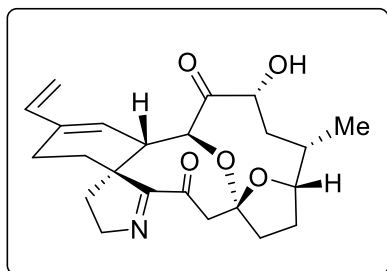


**Portimine B**

**B**

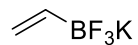


15 – 25



**Portimine B**

15. L-selectride, THF,  $-78\text{ }^{\circ}\text{C}$
16.  $\text{NaBH}_4$ , MeOH,  $0\text{ }^{\circ}\text{C}$
17. TEMPO, NaOCl, KBr,  $\text{NaHCO}_3$ , DCM,  $\text{H}_2\text{O}$ ,  $0\text{ }^{\circ}\text{C}$
18. Zn, AcOH,  $\text{H}_2\text{O}$ ,  $70\text{ }^{\circ}\text{C}$
19. TBSOTf,  $\text{Et}_3\text{N}$ , DCM, reflux
20. DMDO (excess), acetone, DCM,  $0\text{ }^{\circ}\text{C}$
21.  $\text{Ac}_2\text{O}$  (excess),  $\text{Et}_3\text{N}$ , DCM,  $35\text{ }^{\circ}\text{C}$
22. LiOH, THF,  $\text{H}_2\text{O}$ ,  $0\text{ }^{\circ}\text{C}$
23. **6**, Pd(dppf) $\text{Cl}_2$ ,  $\text{Et}_3\text{N}$ , *n*-PrOH,  $90\text{ }^{\circ}\text{C}$
24. DMP,  $\text{NaHCO}_3$ , DCM, r.t.
25.  $\text{NH}_3$ ,  $\text{H}_2\text{O}$ , MeOH



**6**

Step 18: Please, suggest a mechanism.

Hint step 20: Oxidation at 2 positions.

Step 21: Please, show the mechanism.  
Which analogous named rearrangement uses 2-alkylpyridine-*N*-oxides as substrates? Classify the rearrangement.

Hint step 22: Selective mono-deprotection.