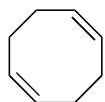
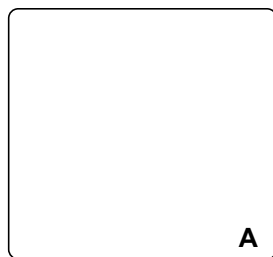


# $\alpha$ -Ketenyl Radical Intermediates in the Synthesis of Propellanes. A Formal Synthesis of Modhephene

Benoit De Boeck and Gerald Pattenden  
*Tet. Lett.* **1988**, 39, 6975 – 6978.



1-4



A

7-12



B

see next page

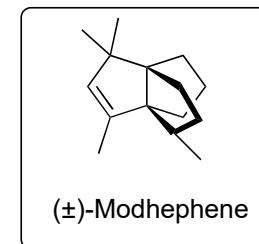
- 1)  $\text{BH}_3 \cdot \text{THF}$ , then  $\text{NaOH}$ ,  $\text{H}_2\text{O}_2$
- 2)  $\text{TBSCl}$  (1 eq), imidazole,  $\text{CH}_2\text{Cl}_2$
- 3)  $\text{TPAP}$  cat.,  $\text{NMO}$ ,  $\text{CH}_2\text{Cl}_2$
- 4)  $\text{TiCl}_3 \cdot \text{AlCl}_3$ ,  $\text{Li}$ , then add product from reaction 3) and acetone,  $\Delta$

- 7) 40%  $\text{HF}$ ,  $\text{H}_2\text{O}-\text{CH}_3\text{CN}$
- 8) Swern
- 9)  $\text{Me}_3\text{SiCH}_2\text{CO}_2\text{Et}$ ,  $\text{LDA}$
- 10) 1M  $\text{NaOH}$
- 11) (*o*-iodophenyl) $\text{CH}_2\text{CH}_2\text{SH}$ ,  $\text{DCC}$ ,  $\text{DMAP}$ ,  $\text{CH}_2\text{Cl}_2$
- 12)  $\text{Bu}_3\text{SnH}$ ,  $\text{AIBN}$ , benzene,  $\Delta$

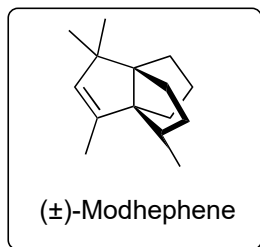
Step 4: Name Reaction? Mechanism?

Step 9: Discuss the mechanism of the reaction, advantages and disadvantages compared to other methods.

Step 12: Mechanism?



13-20



- 13) NaH, MeOCHO
- 14) TsN<sub>3</sub>, Et<sub>2</sub>NH
- 15) CuSO<sub>4</sub>, toluene, Δ
- 16) *t*-BuLi, CO<sub>2</sub>, *then* CH<sub>2</sub>N<sub>2</sub>
- 17) Me<sub>2</sub>CuLi
- 18) MeI, KO*t*-Bu, *then* LiI, collidine
- 19) LiAlH<sub>4</sub>
- 20) Martin's Sulfurane

Step 15: Hint: a new ring is formed in this step

Step 17: Mechanism? Rationalize and explain the synthetic utility of this type of strategy and discuss how it could be extended to other types of reactions.

Step 20: Structure of Martin's Sulfurane? What are other reagents that you could use for this transformations?