

Institutsseminar

Conformationally selected ions react differently with neutrals

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The conformation of a molecule plays an important role in the chemistry of complex molecular systems, necessitating the development of experimental methods for the characterization of the chemical reactivity of individual conformers. Previously, the *cis* and *trans* conformers of neutral 3-aminophenol have been spatially separated in the gas phase based on their different dipole moment and reacted with a 'Coulomb crystal' of calcium ions [1]. These crystals are structures formed by trapped, laser-cooled ions and they can be used to 'sympathetically cool' other trapped ionic species. Building up on these methodologies, ionic Diels-Alder reactions were studied conformationally selected neutral dibromobutadiene (DBB) molecules and sympathetically cooled propene ions. These studies provided new insights into conformationally dependent reaction mechanisms of cycloaddition reactions [2].

In order to gain an even deeper understanding of conformational effects of chemical reactions, conformational control of not only the neutral, but also the ionic reactants in ion-molecule reactions would be beneficial. To this end, we used a two-color (1+1') resonance-enhanced multiphoton ionization (REMPI) scheme [3] to selectively ionize the *cis* and *trans* conformers of *meta*-aminostyrene (mAS) molecules inside the ion trap and sympathetically cool them using calcium Coulomb crystals [4]. Following the developed ion loading scheme, conformationally selected mAS⁺ ions were individually reacted with calcium atoms and with 1,3-cyclohexadiene molecules. Different reactivities of

cis and trans mAS⁺ ions were observed, expanding the scope of conformational selectivity to ionic species and paving the way towards fully conformationally controlled chemistry.

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References

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