Note:

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The legally binding versions are found in the University of Innsbruck Bulletins (in German).

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Curriculum for the

Master's Programme Chemistry
at the Faculty for Chemistry and Pharmacy
of the University of Innsbruck

§ 1 Qualification Profile

- (1) The Master's Programme Chemistry is classified among the natural science studies.
- (2) The Master's Programme Chemistry aims to deliver the training required to qualify chemists by imparting the special subject knowledge, competences and methods of chemical science research and the responsible behaviour expected of their profession. The master's programme is the basis for entry to the profession of chemist in the fields of research, technology, the environment and areas of chemical relevance to the public authorities. Graduates are qualified to carry out autonomous scientific research in chemical subjects and to take on leading positions as well as to make use of their acquired competences on an interdisciplinary basis in order to solve chemical problems.
- (3) The Master's Programme Chemistry lays the foundations for studying for a doctorate in Chemistry or related disciplines.
- (4) The Master's Programme Chemistry focuses on current research fields closely linking theoretical training and experimental practical competences. The Master's Programme Chemistry offers a thematically extensive training in chemistry and at the same time it allows the students to select contents for a far-reaching specialisation corresponding to their interests. Besides of acquiring advanced and up-to-date knowledge in the chemical sub-disciplines, the master's programme also imparts interdisciplinary key competences and a sense of responsibility regarding the benefits and risks of natural science research and applications.

§ 2 Scope and Duration

The Master's Programme Chemistry covers 120 ECTS-Credits. This corresponds to a duration of four semesters. One ECTS-Credit corresponds to a work load of 25 hours.

§ 3 Admission

- (1) Completion of a relevant University Bachelor Programme or a relevant Bachelor Programme at a University of Applied Science, or completion of other equivalent studies at an accredited Austrian or non-Austrian post-secondary educational institution is required for admission to the Master's Programme Chemistry.
- (2) A completed Bachelor Programme Chemistry at the University of Innsbruck is in any case a relevant study programme. The rectorate decides based on the regulations specified in the Universities Act on the admission for graduates having completed other study programmes at approved Austrian or non-Austrian post-secondary education institutions or on the equivalence of the achievements.
- (3) In the event that equivalence has been established in principle but with certain qualifications missing for full equivalence, supplemental examinations may be required. These examinations must be completed during the Master's Programme.

§ 4 Types of courses and maximum number of students per course

- (1) Courses without continuous performance assessment:
 - 1. **Lectures** (VO) are courses held in lecture format. They introduce the research areas, methods and schools of thought for a given subject. No maximum number of participants;
- (2) Courses with continuous performance assessment:
 - 2. **Seminars** (SE) provide in-depth treatment of scientific topics through students' presentations and discussion thereof. Maximum number of participants: 60
 - 3. **Lectures with practical elements** (VU) focus on the practical treatment of concrete scientific tasks that are discussed during the lecture parts of the course. Maximum number of participants: 60
 - 4. **Practical training courses** (PR) provide practical experience with concrete scientific tasks, complementing occupational and ac ademic training. Maximum number of participants: 10

§ 5 Allocation of places in courses with a limited number of participants

In courses with a limited number of participants, course places are allocated as follows:

- 1. Students for whom the study duration would be extended due to the postponement are to be given priority.
- 2. If the criterion no 1 does not suffice, first, students for whom this course is part of a compulsory module are to be given priority, and second, students for whom this course is part of an elective module.
- 3. If the criteria no. 1 and no. 2 do not suffice, the time of the acquisition of the prerequisites for registration will apply.
- 4. If the criteria no. 1, no. 2, and no. 3 do not suffice, the grade of the module, which is directly required for this course, is to be used.
- 5. If the criteria mentioned before do not suffice, the available places are drawn by random.

§ 6 Compulsory and Elective Modules

- (1) The Master's Programme Chemistry is divided into the following groups of modules:
 - 1. Elective modules in the six sub-disciplines of Chemistry: Analytical Chemistry, Inorganic Chemistry, Biochemistry, Organic Chemistry, Physical Chemistry and Theoretical Chemistry. Modules corresponding to 62.5 ECTS-Credits must be passed in five of the sub-disciplines from these elective modules.
 - 2. Elective modules of the advanced studies of the sub-disciplines of Analytical Chemistry, Inorganic Chemistry, Biochemistry, Organic Chemistry, Physical Chemistry and Theoretical Chemistry: Modules corresponding to 15 ECTS-Credits must be passed from these elective modules.
 - 3. General skills elective modules: Modules corresponding to 10 ECTS-Credits must be assed from these modules.
 - 4. Compulsory module: Defence of the Master's Thesis (2.5 ECTS-Credits).
- (2) Elective modules of the sub-disciplines Analytical Chemistry, Inorganic Chemistry, Biochemistry, Organic Chemistry, Physical Chemistry and Theoretical Chemistry:

From the following elective modules, modules in five sub-disciplines of Chemistry corresponding to 62.5 ECTS-Credits must be passed:

1.	Elective Module: Analytical Chemistry A	h	ECTS- Credits
a.	VO Basics and Application of Modern Separation Techniques Classic separation techniques (filtration, centrifugation, dialysis, distillation, liquid-liquid and solid phase extraction), electrophoresis (zone electrophoresis, 2D gel electrophoresis, isotachophoresis, isolectric focusing), chromatic chromatography (TLC, GC, GPC and HPLC) and preparative colum chromotography (basics, phase systems, instrumentation, examples)	2	2
b.	VO Bioanalytics and Coupling Methods Separation methods for biomolecules (chromotography, electrophoresis), structural analysis of biomolecules, coupling with mass spectrometry and nuclear resonance spectroscopy	1	1.5
c.	VO Methods of Spectroscopic Analysis Fundamentals and theory of UV-, MIR-, NIR and Raman spectroscopy, examples of applications in industry and research	1	1.5
d.	VO Modern Applications of Electroanalytic Techniques Functioning of redox mechanisms, trace analysis, corrosion tests, selective and sensitive detection methods for chromatography and capillary electrophoresis	1	1.5
e.	VO Trends in Separation Techniques Stationary phases for chromatography of liquids (synthesis, characterisation, selection and optimisation of methods), special detection methods; miniaturisation of separation techniques; e.g. chip technologies for electrophoresis and chromatography	1	1
	Total	6	7.5
	Learning Outcomes: Students of this module gain a profound knowledge of all current methods of analysis. acquire the skill to independently apply this knowledge for solving actual problem consideration of the strengths and limits of application of the respective methods. Prerequisites: none		

2.	Elective Module: Analytical Chemistry B	h	ECTS- Credits
	PR Advanced Practical Course in Instrumental Analysis Literature research and selection of suitable methods for a given analytical problem, selected practical examples from the areas of environmental, food, bio, polymer and industrial analytics in application of electrophoretic, chromatographic, electrochemical, atomic-spectroscopic and molecule-spectroscopic as well as coupling analysis methods, preparation of actual samples, evaluation of data and comparison of methods	5	5
	Total	5	5
	Learning Outcomes: Students are able to select tailored methods of analysis for specific problems interpret the results. Prerequisites: successful completion of elective module 1	and eva	luate and

3.	Elective Module: Inorganic Chemistry A	h	ECTS- Credits	
a.	VO Solid State Chemistry II Introduction to "inorganic functional materials" relevant to material science with main focus on hard materials, alloys and nanoscale materials; Besides of synthesis, the technical relevant electronic, optic and magnetic characteristics of these material are focused on.	2	3	
b.	VO Organometallic Chemistry and Homogeneous Catalysis Classification, production, structure, stability, stoechiometric and catalytic reactivity, applications of organometallic compounds in research and industrially relevant processes as well as current developments and challenges of organometallic chemistry	3	4.5	
	Total	5	7.5	
	Learning Outcomes: Students gain advanced knowledge and competences in inorganic cher consideration of actual application of inorganic functional materials in industry. Students understand the manifold relations between structure and inorganic materials. Prerequisites: none	in technology and		

4.	Elective Module: Inorganic Chemistry B	h	ECTS- Credits
	PR Advanced Practical Course in Inorganic Chemistry Independent experimental working on current topics in research in an inorganic chemistry working group; actual application of advanced methods of synthesis and spectroscopic and diffractrometic characterisation of materials: main focus either on organometallic chemistry and catalysis, coordination chemistry, magnetochemistry, photochemistry, material science or solid state chemistry	5	5
	Total	5	5
	Learning Outcomes: Students are able to design location, function and characteristics of molecular solid bodies by application of the wide range of methods of synthesis of in an application-orientated fashion.		
	Prerequisites: none	•	

5.	Elective Module: Biochemistry	h	ECTS- Credits
a.	VO Advanced Biochemistry I In-depth study or structure and function of proteins, especially: chemistry of amino acids building blocks; peptide bonds; protein analysis; conformation, folding, decomposition and dynamic function of proteins; allosteric proteins; mechanisms of enzymatic catalysis; protein sequence motifs (bioinformatics)	2	3
b.	VO Advanced Biochemistry In-depth study of biochemical regulation and signal transduction processes, especially: amino acid metabolism, cholesterol metabolism, steroid hormones, isoprenoid compounds, chemical attributes of DNA, gen-protein relation, gen regulation, protein targeting, mitogenic signal transduction, molecular basics of carcinogenesis	2	3
c.	VO Introduction to the Advanced Practical Course in Biochemistry Theoretical basics of modern methods in biochemistry and genetic engineering, possible applications in biochemical fundamental research and in medicine and pharmacology	1	1.5
d.	PR Advanced Practical Course in Biochemistry Research-orientated practical training in modern methods of biochemistry and genetic engineering, especially: recombinant protein expression, protein purification, protein-DNA-interactions, analysis of gene expression, gene transfer, cell transformation	5	5
	Total	10	12.5
	Learning Outcome: Students acquire in-depth knowledge of objects and methods in research in genetic engineering. Main focus is on in-depth study of chemical characteristics of nucleic acids and proteins, the most important biomolecuthe flow of the genetic information. Knowledge for applying methods of genetic engineering for issues in medicine are also imparted, especially basics of physiological and pathophysiological processes and the application biological approaches for determining the functioning of biomolecules.	al and biological cules in relation to f biochemistry and on the molecular	
		uon oi	

6.	Elective Module: Organic Chemistry A	h	ECTS- Credits
a.	WO Advanced Organic Synthesis Modern synthesis concepts (e.g. atom-economic synthesis, biomimetic synthesis, green synthesis chemistry, etc.): synthesis strategies (e.g. linear vs. block construction, solution vs. solid phase synthesis, use of natural materials (from the chiral pool) and stereo-selective synthesis, pericyclic reactions, biocatalysis, etc.) and synthesis methods (thermo-, photo-, electro-synthesis, synthesis with organometallic complexes and radicals, protective group techniques, etc.) as well as total syntheses of natural materials	2	3
b.	VO Bioorganic Chemistry Fundamentals of bioorganic chemistry; organic-chemical synthesis as approach for natural analogues, which leads to the deliberate manipulation of the properties of biological systems; structural basis of biocatalysis and special stereo-chemical aspects	2	3

Learning Outcomes: Students acquire advanced knowledge and skills in organic chemistry in special consider of actual application of modern organic synthesis, bioorganic chemistry and other organic chemistry and other organic synthesis.	c.	SE Seminar in Biologic Organic Chemistry Discussion and presentation of current topics in organic chemistry with focus "structure, reactivity & synthesis" and chemical-biological approaches; training programme for dealing with primary literature and discussion of lectures	1	1.5	
Students acquire advanced knowledge and skills in organic chemistry in special consider of actual application of modern organic synthesis, bioorganic chemistry and other organic		Total	5	7.5	
research fields of organic chemistry, perfecting of the presentation technique. Prerequisites: none		Students acquire advanced knowledge and skills in organic chemistry in special consideration of actual application of modern organic synthesis, bioorganic chemistry and other organic chemical approaches for molecular, biological problems. Independent dealing with current research fields of organic chemistry, perfecting of the presentation technique.			

Elective Module: Organic Chemistry B	h	ECTS- Credits
PR Advanced Practical Course in Organic Chemistry Practical implementation of organic-chemical synthesis steps by using modern strategies and methods for making selective chemical conversions. <i>Organisation</i> : rotation principle, leading through current research topics in organic chemistry; actual application of advanced synthesis methodology and spectro-analytical characterisation of materials	5	5
Total	5	5
Learning Outcomes: Students are able to make experiments independently. They are able to dissues as well as with complex problems and questions of basic research.	leal with	n practical
	le	al with

8.	Elective Module: Physical Chemistry	h	ECTS- Credits
a.	VU Properties of Solid Materials Structure, lattice vibrations, electronic properties, metals, isolators, semiconductors	2	2
b.	VO Kinetics and Catalysis Complex reaction kinetics, non-linear and oscillating systems, microscopic basics of kinetics	2	3
c.	SE Current Topics in Material Science and Physical Chemistry New materials, sustainable energy systems, surface and interface phenomena, modern methods in physical chemistry	2	2.5
d.	WO Atmospheric Chemistry Basics of atmospheric chemistry; layers of the atmosphere; chemistry of the troposphere (hydrologic cycle, clouds, pollutant balance, smog, degradation mechanisms, half-life periods); chemistry of the ozone layer (stratosphere), chemistry of higher atmospheric layers as well as of the interstellar area; history (and future?) of the atmosphere; extra-terrestrial atmospheres; greenhouse effect; "indoor chemistry"	1	1.5
e.	PR Experiments in Applied Physical Chemistry e.g. fuel cell, electrolysis cell, mass spectrometry, quantitative gas analysis	2	2.5
f.	PR Thin Film Technology, Chemical Vapour Deposition Calculation and production of functional thin film systems with chemical vapour deposition	1	1
	Total	10	12.5

Learning Outcomes:

Students understand the interrelations between solid body structure, possible stimuli for this structure and the macroscopic material properties. They are able to describe complex reactions of real processes and explain the microscopic fundamentals. Students are able to present the most important processes of atmospheric and environmental chemistry. They independently deal with current research fields in physical chemistry relevant for society and perfect their presentation technique. Students carry out experimental work on application-orientated, complex problems independently. They calculate functional thin layer systems, different deposition techniques and check the material properties of the product made by them.

Prerequisites: none

9.	Elective Module: Theoretical Chemistry	h	ECTS- Credits
а.	VO Advanced Methods in Quantum Chemistry Ab initio quantum chemistry, Hartree-Fock methods, post-Hartree-Fockmethods, density functional theory, perturbation theory, energy hypersurfaces, application examples	2	3
b.	VO Simulation Methods Molecular dynamics simulations, hybrid methods (quantum mechanics + molecular mechanics), free energy calculations, Monte-Carlo-simulations	2	3
c.	VO Molecular Modelling Chemoinformatics, molecular descriptors, similarity, virtual screening, structure-based design, chemical databases	1	1.5
d.	PR Advanced Exercises in Theoretical Chemistry and Computational Chemistry Practical application of calculation methods of the master's programme	5	5
	Total	10	12.5
	Learning Outcomes: Students are able to apply computational chemical methods for modelling molecules and materials. They use classic forces and forces of quantum-mechanics for describing dynamic processes and for predicting thermodynamic properties. Based on examples of scientific practice, they apply MC and MD methods, advanced calculation methods of quantum mechanics as well as methods dealt with in the lectures Prerequisites: none		

(3) Elective modules in advanced studies in the chemical sub-disciplines of Analytic Chemistry, Inorganic Chemistry, Biochemistry, Organic Chemistry, Physical Chemistry and Theoretical Chemistry

Modules corresponding to 15 ECTS-Credits must be passed from the following elective modules:

10.	Elective Module: Advanced Study of Analytical Chemistry A	h	ECTS- Credits
a.	VO Analytical Methods for Materials Traditional methods; porosimetry, , BET, RFA; new analytical methods for materials: infrared and Raman spectroscopy	1	1.5
b.	VO Sensor Technology Structure, basic functionality of different types of sensors; application of electro-chemical, optical sensors, semi-conductor gas sensors, biosensors; modern developments and miniaturisation based on field-effect transistors and sensor arrays	1	1
	Total	2	2.5

Learning Outcomes:

Students acquire skills in determining material properties, in different methods of sensor technology and state-of-the-art analysis methods for some specialist fields.

Prerequisites: none

11.	Elective Module: Advanced Study of Analytical Chemistry B	h	ECTS- Credits
a.	VO Food Analysis Environmental-analytical concepts, inspection systems and checkpoints as well as their interpretation	1	1.5
b.	VO Laboratory Diagnostic Methods of Analysis Sample taking, analytics and diagnostics of biological samples (blood, urine, liquor), molecular-biological methods of analysis (PCR diagnostics, mutation diagnostics, ELISA method), immunological methods (enzyme immunoassays – EIA), biomarker-analytics (MALDI, SELDI, MELDI) in the field of genomics, proteomics and metabolomics	1	1.5
c.	VO Drug Analysis Methods of extraction of natural materials (e.g. microwave extraction, PLE, SFE) and sample purification (solid phase extraction, LLE); separation of natural materials with special focus on coupling with mass spectrometry	1	2
	Total	3	5
	Learning Outcomes: Students acquire knowledge of current trends in separation techniques and in the applic of modern analysis methods for analysing food and phytochemical in laboratory diagnoand industry.		
	Prerequisites: none		

12.	Elective Module: Advanced Study of Analytical Chemistry C	h	ECTS-Credits
a.	VO Chemometric Methods in Analytical Chemistry Variance analysis, multivariate data analysis (main component analysis, cluster analysis), multivariate regression methods (MLR, PCR, PLS), statistical planning of tests (screening designs, optimisation designs, mixing designs)	2	3.5
b.	VO Industrial Analysis Finding of solutions for analytical problems of industry, requirements of the resp. analysis methods, establishment preconditions, evaluation and validation of analysis methods, measurement system analysis (MSA), fundamentals of quality control	1	1.5
	Total	3	5
	Learning Outcomes: Students acquire knowledge in advanced data analysis, evaluation and calibration method strategies for planning experiments and fundamentals of quality control and approaches the finding solutions to analytical problems of research, development and industry.		
	Prerequisites: none		

13.	Elective Module: Advanced Study of Inorganic Chemistry A	h	ECTS- Credits	
a.	VO Advanced Solid State Chemistry Advanced study of solid state chemistry in special consideration of modern synthesis strategies such as high temperature and high pressure syntheses; insights into modern solid-state-specific methods of characterisation and introduction to current research fields and applications in solid state chemistry	1	2	
b.	PR Practical Course in Applied High-Pressure Solid State Chemistry Experimental conducting of modern high pressure syntheses (multianviltechnology) with focus on current issues in synthesis of new functional materials	2	3	
	Total	3	5	
	Learning Outcomes: Students are familiar with current research fields in modern solid state chemistry. The acquire advanced practical skills for making functional materials with high pressure synthesis. Prerequisites: none			

14.	Elective Module: Advanced Study of Inorganic Chemistry B	h	ECTS- Credits
a.	VO Current Topics in Inorganic Chemistry Design, functioning and current applications of inorganic materials	1	2
b.	VO Photochemical Transformation of Solar Energy Principles, applications and state of the technology for photochemical winning of chemical energy from solar energy by using supramolecular systems; production of photoelectric energy from solar energy (photovoltaic, Grätzel cell); comparison to natural systems of photosynthesis within the scope of coordination chemistry and Marcus theory	1	1.5
c.	VO Heteronuclear-NMR-Spectroscopy Fundamental relations between nuclear spin, magnetic dipole moment, Larmor frequency and Zeeman energy; assessment of signal intensity; referencing of spectra; effects of indirect spin-spin-coupling	1	1.5
	Total	3	5
	Learning Outcomes: Students acquire advanced knowledge for applying inorganic materials in fields of materials science. Students are familiar with the methods of structure by using metal/heteronuclear-NMR-spectroscopy. Prerequisites: none		

15.	Elective Module: Advanced Study of Inorganic Chemistry C	h	ECTS- Credits
a.	VO X-Ray Diffraction on Single Crystals Principles, methods, characteristics and state of technology in X-ray structure analysis of single crystals	1	1
b.	PR Practical Course on Diffraction Methods Methods of single crystal X-ray structural analysis, independent execution of single crystal structure analyses of selected coordination compounds, organometallic compounds and solid state materials, interpretation and computer-assisted assessment and visualisation of data and the structural characterisation of inorganic materials in the solid state	2	1.5
	Total	3	2.5

Learning Outcomes:
Students acquire theoretical and practical skills for methods and applications of X-ray
defraction on single crystals.
Prerequisites: none

16.	Elective Module: Advanced Study of Biochemistry A	h	ECTS- Credits
a.	VO Advanced Biochemistry III Regulation of gene expression, gene silencing, gene mutation, DNA-	1	2
	microarray technology, gene isolation, gene transfer, gene therapy		_
b.	PR Advanced Practical Course in Biochemistry Training in modern biochemical and genetic-technological methods for isolating, structural characterisation and functional analysis of specific target genes and their protein products	2	3
	Total	3	5
	Learning Outcomes: Students acquire in-depth knowledge of modern methods of genetic enginee to their applications in basic research and medicine.	ering wit	h regards
	Prerequisites: none		

17.	Elective Module: Advanced Study of Biochemistry B	h	ECTS- Credits
a.	VO Advanced Biochemistry IV	2	2.5
	Current topics in biochemical research, part 1		2.5
b.	VO Advanced Biochemistry V	2	2.5
	Current topics in biochemical research, part 2	2	2.3
	Total	4	5
	Learning Outcomes:		
	Students acquire in-depth knowledge of current research fields in biochemist	try.	
	Prerequisites: none		

18.	Elective Module: Advanced Study of Organic Chemistry A	h	ECTS- Credits
a.	VO Catalysis of Organic Reactions Introduction to the catalysis of organic reactions, energetic fundamentals; acid base catalysis versus transition metal catalysis versus biocatalysis; heterogeneous versus homogeneous catalysis with reference to solid phase synthesis of organic compounds, catalysis with proteins and nucleic acids – current problems	1	1.5
b.	VO Mechanisms of Organic Reactions Mechanistic fundamentals and methods (isotope effects, linear relations, state correlations etc.), influence of the medium, molecular activation, stepwise versus synchronous reactions (reactive intermediates; pericyclic reactions), current problems.	1	1.5
c.	VO Stereochemistry & Supramolecular Chemistry Systematics of stereochemistry, symmetry and symmetry elements in (larger) organic compounds, organisation principles and functions of supramolecules, use of supramolecular compounds in chemical-biological synthesis based on current examples	1	2
	Total	3	5

Learning Outcomes:

Students acquire advanced knowledge on the reactivity of organic compounds in current chemical, chemical-biologic and nonchemical research fields. Students get familiar with the analysis of reaction channels and are able to apply modern concepts in synthesis planning (from simple chemical bonds to biomolecules and polymer materials).

Prerequisites: none

19.	Elective Module: Advanced Study of Organic Chemistry B	h	ECTS- Credits
a.	VO Organic Structural Chemistry I		
	Spectroscopic characterisation of organic compounds, nano materials and	1	1.5
	biomolecules using NMR spectroscopy		
b.	VO Organic Structural Chemistry II		
	Spectrometric characterisation of organic compounds, nano materials and	1	1.5
	biomolecules using modern mass spectrometric methods		
c.	PR Practical Course in Organic Structural Chemistry		
	Characterisation of a synthesis product or a natural material using	3	2.
	(heteronuclear) NMR-spectroscopy, mass spectrometry, UV-VIS-, CD-,	3	2
	IR- and fluorescence spectroscopy		
	Total	5	5
	Learning Outcome:		
	Students acquire advanced knowledge in structure analysis of organic com-	pounds i	n current
	chemical, chemical-biologic and nano-chemical research fields. Students	are able	to apply
	modern methods of structure analysis of low-molecular chemical compound	s of bion	nolecules
	and polymer materials.		
	Prerequisites: none		

20.	Elective Module: Advanced Study of Organic Chemistry C	h	ECTS- Credits
	PR Solid Phase Synthesis & Natural Materials Isolation Independent experimental work in an organic chemistry working group on current topics in research; practical execution of an automated solid phase synthesis, the isolation of a natural material or of a targeted natural material transformation or synthesis	2	2.5
	Total	2	2.5
	Learning Objectives: Students acquire advanced laboratory skills in modern methods of solid planatural material isolation, transformation or synthesis.	nase syn	thesis, of
	Prerequisites: none		

21.	Elective Module: Advanced Study of Physical Chemistry A	h	ECTS- Credits
a.	VU Kinetics and Dynamics of Surface Reactions Mechanisms of molecular and dissociative adsorption. Potential energy surface, activated adsorption, nuclear and molecular coupling on surfaces in MO-picture, band theory and density of states, catalysis, volcano relationship	1	1
b.	VU Energy Engineering and Catalysis Electronic and structural principles of heterogenous catalysis, physical- chemical properties of nanostructured materials, environmental catalysis, emission control, processes for chemically storing and converting energy, carbon storage and use	1	1

c.	VU Electrochemistry and Applications in Energy Research		
	In-depth study of examination methods (.e.g. rotating (ring) disk electrode,		
	electrochemical impedance spectroscopy), semiconductor electrochemistry	1	1
	(e.g. Mott-Schottky analysis), fundamentals of elctroanalysis and li-ion		
	insertion and application (e.g. in fuel cells or li-ion batteries)		
d.	PR Current Research in Physical Chemistry		
	Current research work in physical chemistry working groups selected by		
	the students, e.g. characterisation and structure examination of surface-		
	and adsorbtion systems (LEED, STM, ARUPS), catalytic carbon	2	2
	hydrogenation for energy carriers, conversion of energy carriers for	2	2
	hydrogen production, product analysis ((MS, GC), catalytic		
	characterisation of SOFC-relevant materials, electroanalysis and battery		
	research, photoelectric chemistry		
	Total	5	5
	Learning Outcomes:		
	Students acquire advanced knowledge and an understanding of the r	elations	between
	electronic and geometric structures of surfaces and the kinetics and dyn	amics o	f surface
	reactions. They get advanced knowledge of surface and nano sciences in r	elation t	o energy
	and environmental technology and of heterogenous catalysis (environ	mental	catalysis,
	chemical energy storing and conversion, carbon storing and use). Students le		•
	electrochemical thermodynamics and kinetics with focus on applications in		
	e.g. batteries, fuel cells an solar cells.	23	,
	Prerequisites: none		

22.	Elective Module: Advanced Study of Physical Chemistry B	h	ECTS- Credits
a.	VU Materials in Cryo-Conditions Fundamentals of cryochemistry, especially of aqueous solutions or volatile components resp.; freezing and defrosting properties; freeze concentration; glazing, cold crystallization; application in astronomy (forming of planets, stars, galaxies from interstellar dust; chemistry of comets); atmosphere chemistry (ice clouds), geology (glaciers and ice sheets), biology (cryomicroscopy) and medicine (cryonics) as well as technology (defrosting methods, technical snow) and food industry (freeze-drying)	1	1.5
b.	PR Practical Course on Materials in Cryo-Conditions Working with current research methods, e.g. making and analysis of aqueous solutions in cryo-conditions, analysis especially based on cryo-microscopy, cryo-XRS and cryo-calorimetry; production using vitrification, chemical vapour deposition or high-pressure cryosynthesis	1	1
	Total	2	2.5
	Learning Outcomes: Students acquire knowledge on the behaviour of materials in cryo-co experimentally regulating material properties. Prerequisites: none	nditions	and for

23.	Elective Module: Advanced Study of Physical Chemistry C	h	ECTS- Credits
a.	VO Surface and Material Analysis Methods for determining the structure and chemical composition of surfaces, interfaces and layered systems:: AES, XPS, depth profile analysis and adsorption spectroscopy	1	1.5
b.	PR Practical Course in Surface and Material Analysis Working with current research methods, e.g. surface and depth-profile analysis using X-ray photoelectron spectroscopy (XPS), adsorption spectroscopy	1	1
c.	VO Scanning Probe and Electron Microscopy Principles and operation modes of scanning probe microscopy, atomic force microscopy, surface-potential microscopy, electric-force-microscopy, friction microscopy and transmission electron microscopy	1	1.5
d.	PR Practical Course in Scanning Probe and Electron Microscopy Characterisation of surfaces in the nanoscopic area and resolution by using scanning problem methods, examinations of nano-particles and layers materials with transmission electron microscopy	2	1
	Total	5	5
	Learning Outcomes: Students acquire well-founded theoretical and practical knowledge of modern meth surface and material analysis for dealing with technically-relevant problems. Prerequisites: none		ethods of

24.	Elective Module: Advanced Study of Theoretical Chemistry A	h	ECTS- Credits
a.	VO Theoretical Treatment of Biomolecules Bioinformatics, biomolecular databases, sequence analysis, prediction of	2	2.5
	RNA/DNA structures, prediction of protein folding and protein structures, DNA-, RNA- and protein dynamics	4	2.3
b.	PR Theoretical Treatment of Biomolecules		
	Application of methods for describing biological macromolecules and their	2	2.5
	interactions		
	Total	4	5
	Learning Outcomes:		
	Students acquire theoretical and practical knowledge for applying modern	n techno	logies of
	DNA-, RNA- and protein sequence analysis, for predicting protein and nucleic acid structures		
	and for predicting biomolecular dynamics.		
	Prerequisites: none		

25.	Elective Module: Advanced Study of Theoretical Chemistry B	h	ECTS- Credits
а.	VO Introduction to Computer-Assisted Material Science Polarizability and multibody effects, reactive force fields, periodic approaches in quantum mechanics, density functional theory, application examples	2	2.5
b.	PR Numerical Methods – Computer Operations for Determining Physical-Chemical Properties Handling of various codes for numerical calculation of material properties	2	2.5
	Total	4	5
	Learning Outcome: Students gain the competence and practical experience for dealing with modern numerical methods for calculating material properties. Prerequisites: none		

(4) Elective Modules General Skills

Modules amounting to $10\,\mathrm{ECTS}\text{-}\mathrm{Credits}$ must be passed from the following elective modules:

26.	Elective Module: Intellectual Property Rights and Legal Framework of Chemistry	h	ECTS- Credits
	VO Intellectual Property Rights and Legal Framework of Chemistry:		
	Patent and Chemicals Law		
	Copyright, trademark law, patent law, Austrian and European chemicals	2	2.5
	law, assessment and licence authorisation of chemicals and active		
	ingredients		
	Total	2	2.5
	Learning Outcomes:		
	Students acquire an understanding of the intellectual property rights relev	ant for	chemical
	scientists and get an overview of the legal regulations in relation to chemical	s.	
	Prerequisites: none		

27.	Elective Module: Project Management	h	ECTS- Credits
	VU Project Management Project definition, project management approaches and processes, practice-orientated tools for planning, organising, implementing and controlling projects; chemistry-relevant examples from the fields of research and industry	2	2.5
	Total	2	2.5
	Learning Outcomes: Students understand the importance, methodology and factors of success of modern project management and learn to apply management processes and helpful tools for their own projects. The acquired competences make it possible for the students to assume an active role in project organisation		
	Prerequisites: none		

28.	Elective Module: Lecture Series GÖCh/CMBI/ Material and Nano Science	h	ECTS- Credits
	SE Lecture Series GÖCh/CMBI/Material and Nano Science Participation in lectures of invited guests within the scope of the series of the Society of Austrian Chemists (GÖCh) and/or the Centre for Molecular Bio Sciences in Innsbruck (CMBI) and/or the focus in Material and Nano	2	2.5
	Science. Total	2	2.5
	Learning Outcomes: Through participation in the lecture series students are familiarised with the current research topics of external experts and learn how current topics are presented and discussed on a scientific level. Contact with the invited professors allows students to get to know the Scientific Community.		
	Prerequisites: none		

29.	Elective Module: Interdisciplinary Skills	h	ECTS- Credits
	Courses from the offer of the master's programmes at the University of Innsbruck or from the field of "equality and gender" amounting to 5 ECTS-Credits are to be chosen.		5
	Total		5

	Learning Outcomes:
	Advanced qualification as chosen by the student.
	Prerequisites: The registration requirements specified by the respective curricula must be
	met.

30.	Elective Module: Computer-Aided Database Research	h	ECTS- Credits
	VU Computer-Aided Database Research Structure and information content of chemical-science databases (SciFinder, Beilstein Reaxys, Science of Synthesis – Houben Weyl, esp@cenet, Cambridge Crystallographic Data Centre etc.); strategies of literature research, search algorithms and search profiles, data management	2	2.5
	Total	2	2.5
	Learning Outcomes: Students acquire application-orientated knowledge of the information of information search in chemistry-related databases.	contents	and the
			103

31.	Elective Module: Measurement Technology and Computer-Assisted Control of Experiments	h	ECTS- Credits
	PR Measurement Technology and Computer-Assisted Control of		
	Experiments	3	2.5
	Measurement technology, e.g. basic components of A/D- and D/A	3	2.3
	conversion, programming in LABVIEW		
	Total	3	2.5
	Learning Outcomes:		
	Participants get to know hard- and software (programming) for capturing r	neasurei	nent data
	and controlling experiments.		
	Prerequisites: none		

32.	Elective Module: Metal and Ceramics Processing for Laboratory Use	h	ECTS- Credits
	PR Metal and Ceramics Processing for Laboratory Use	5	5
	Independent work in the precision-mechanic workshop	۲	3
	Total	5	5
	Learning Objectives: Participants learn methods or acquire skills resp. of precision mechanics and are able		
			e able to
	independently make their own mechanic precision elements and apparatuses		
	Prerequisites: none		

33.	Elective Module: Glass Processing for Laboratory Use	h	ECTS- Credits			
	PR Glass Processing for Laboratory Use Independent exercising of glass-blowing and making of glass equipment needed in the laboratory	5	5			
	Total	5	5			
	Learning Outcomes: Participants learn methods of glass processing and are able to independently make glass equipment for laboratories. Prerequisites: none					

(5) Compulsory module: Master's Thesis Defense (2.5 ECTS-Credits):

34.	Compulsory Module: Master's Thesis Defense	h	ECTS- Credits		
	Presentation and defense of one's Master's Thesis (Defensio) within the scope of a 20-minute long scientific presentation with subsequent scientific discussion and questioning by an examination board.		2.5		
	Total		2.5		
	Learning Outcomes: Students are able to present the results of their Master's Thesis in a scientific lecture and are able to defend it.				
	Prerequisites: successful completion of all required modules and the Master's Thesis				

§ 7 Master's Thesis

- (1) Within the scope of the Master's Programme a Master's Thesis corresponding to 30 ECTS-Credits must be written. The Master's Thesis is a scientific work that testifies to the student's ability to work on a scientific topic independently and adequately with regards to content and methodology.
- (2) The topic of the Master's Thesis can be chosen from the areas of Analytical Chemistry, Inorganic Chemistry, Biochemistry, Organic Chemistry, Physical Chemistry or Theoretical Chemistry. Before announcing the topic of the Master's Thesis, the student must provide evidence of having passed a minimum of 60 ECTS-AP in the elective modules.
- (3) Master's Theses are to be presented in the written and electronic forms as specified by the Director of Studies.

§ 8 Examination regulations

- (1) Modules are assessed by module examinations. Module examinations are examinations that serve to show the knowledge and skills gained in a module. The module is completed by positive evaluation of all parts of a module examinations.
- (2) Courses of modules are assessed by course examinations. Course examinations are
 - 1. Examinations that assess the knowledge and skills covered by an individual course and where the performance is assessed by a single examination at the end of the course. The course lecturer must communicate the examination method (written or oral) before the start of the course.
 - 2. Courses with continuous performance assessment, for which the performance assessment is based on regular written and/or oral contributions by the participants.
 - 3. Course lecturers must inform the students on the targets, contents and methods of their course in a suitable fashion before the start of each semester, as well as on the contents, the methods and the evaluations criteria and standards of the course examinations.
 - 4. The module "Master's Thesis Defense" is evaluated by an oral examination before an examination board consisting of three examiners.

§ 9 Academic Degree

Students successfully completing the Master's Programme Chemistry are awarded the title of "Master of Science", shortened to "MSc".

§ 10 Coming into force

- (1) This curriculum comes into force on 1st October 2009.
- (2) The changes of the curriculum as published in the University of Innsbruck Bulletin of 9 May 2016, Issue 26, No. 396 come into force on 1 October 2016 area are to be applied to all students.

Attachment 1: Recommended study plan

As a result of the wide range of electives for the Master's Programme Chemistry, a detailed study plan with a list of courses chosen is not very useful. The time sequence for the chosen subjects is in line with the semester course offerings (winter or summer semester) and the actual choice of modules is decided by the students themselves. Below is an overview plan in line with the relative workload in terms of ECTS-AP:

1st Semester (Winter	2nd Semester	3rd Semester (Winter	4th Semester		
Semester)	(Summer Semester)	Semester)	(Summer Semester)		
Elective N					
(Choice from Modules 1 – 6				
Elective N					
Elective N	Master's Thesis				
	(12.5 ECTS-Credits)		(30 ECTS-Credits)		
(Choice from Modules 1 – 6	5			
Elective N	Defense of the				
	(12.5 ECTS-Credits) Choice from modules 1 – 6		Master's Thesis		
	(2.5 ECTS-Credits)				
Elective Module Sp					
	Module 30				
Se					
	(15 ECTS-Credits)				
S					
	(10 ECTS-Credits)				
C	hoice from modules $21 - 2$	29			